

## Practice test mark scheme Foundation

Q		Expected answers	Marks	Additional guidance
1	a	CO <sub>2</sub>	1	
	b	ionic substances are solids but covalent substances are gases at 20°C	1	
	c	B	1	
	d	nitrogen oxygen	1 1	
	<b>5</b>			
2		added – burning fossil fuels, heating, decay, respiration removed – photosynthesis, dissolving in oceans, chemical reactions	3	Any two  Any two  [4 ways: (3) / 3 ways: (2) / 2 ways: (1)]
			<b>3</b>	
3	a	CO	1	
	b	Zn	1	
	c	C	1	
	d	It is too reactive	1	
	<b>4</b>			
4	a	the outer rigid layer of the Earth	1	
	b	silicon and oxygen	1	
	c	SiO <sub>2</sub>	1	
	<b>3</b>			
5	a	electrolysis	1	
	b	ions cannot move in a solid	1	
	<b>2</b>			
6		<ul style="list-style-type: none"> <li>oxygen and nitrogen have lower boiling points than water</li> <li>oxygen and nitrogen have weak forces between their molecules</li> <li>and so less energy is needed to separate them (and hence lower boiling point)</li> <li>water molecules must have stronger forces between their molecules</li> <li>and so more energy is needed to separate them (and hence higher boiling point)</li> </ul> Any two points but to gain 3 marks, the answer must refer to water and either oxygen or nitrogen	3	Accept 'oxygen and nitrogen are small molecules'
			<b>3</b>	

**TOTAL**      20

Q		Expected answers	Marks	Additional guidance
1	a	electrolysis	1	
	b	from the negative electrode /cathode	1	
	c	$2\text{O}^{2-} \rightarrow \text{O}_2 + 4\text{e}^-$	1	
			<b>3</b>	
2	a	<ul style="list-style-type: none"> <li>oxygen and nitrogen have lower boiling points than water</li> <li>oxygen and nitrogen have weak forces between their molecules</li> <li>and so less energy is needed to separate them (and hence lower boiling point)</li> <li>water molecules must have stronger forces between their molecules</li> <li>and so more energy is needed to separate them (and hence higher boiling point)</li> </ul>	3	Any 3 accept 'oxygen and nitrogen are small molecules'
	b	water	1	
			<b>4</b>	
3	a	the <u>outer rigid layer</u> of the <u>Earth</u>	1	all needed
	b	$\text{SiO}_2$	1	
	c	there are strong (covalent) bonds <u>throughout</u> the structure	1	
			<b>3</b>	
4	a	hydrogen, oxygen, carbon, nitrogen	1	all needed in any order any correct representation without the bonds
	b	$\text{HOOCCH}_2\text{NH}_2 / \text{C}_2\text{H}_5\text{O}_2\text{N}$	1	
	c	carbohydrates do not contain nitrogen (atoms)/amino acids contain nitrogen	1	
			<b>3</b>	
5	a	it has lost oxygen	1	
	b	6.5 (tonnes)	1	
	c	iron / copper or any other metal below carbon in the reactivity series	1	
	d	positive ions in a sea of freely moving electrons	1	
			<b>4</b>	
6		ionic: attraction(1);between oppositely charged ions (1) covalent: attraction between the nuclei of atoms (1); and the electrons shared between them (1)	3	any three
			<b>3</b>	

TOTAL 20